### POSSIBLE OH SUBMILLIMETER EMISSION IN ORION - KL

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RESUMEN. El ión molecular OH no fue detectado en nubes interestelares hasta ahora, debido a que la posible detección de OH en Orión-KL sería una confirmación independiente de las ondas de choque moleculares con o sin campos magnéticos (MHD y disociativas), es estudiada la estructura rotacional fina e hiperfina de este ión. El propósito de la comunicación es dar valores de probabilidades de transición para emisión espontánea e intensidades para frecuencias infrarojas medidas en laboratorio, los cuales deben hacer más fácil la tarea observacional de búsqueda de ésta especie iónica.

ABSTRACT: The OH<sup>+</sup> molecular ion has not been detected in interstellar clouds so far. Since a possible detection of OH<sup>+</sup> in Orion-KL would be an independent confirmation of molecular shock waves with or without the presence of magnetic fields (MHD and dissociative), the fine and hyperfine rotational structure of this ion is studied. It is the purpose of the present communication to provide values of transition probabilities for spontaneous emission and intensities for laboratory measured infrared frequencies which should make easier the observational task to search for this ionic specie.

Key words: INTERSTELLAR-MOLECULES - TRANSITION PROBABILITIES

## . INTRODUCTION

Few molecular species have been detected in interstellar clouds so far, through heir transitions in emission in the far infrared (FIR) and submillimetric (SMM) regions of the lectromagnetic spectrum (Viscuso et al., 1985a,b; Watson et al., 1985a; Melnick,1986; Ziurys, 988). A detailed study of these hyperfine transitions may reveal details of the physical onditions in regions where these molecules are located. In particular in Orion-KL, the nearest mong the giant molecular clouds, which provides an unique source of heterogeneity in physical onditions and chemical abundances within a relatively small spatial region and where many of he chemical differences among, the sub-sources seem to be directly related to processes of tar formation generating shock fronts. Orion-KL emits strongly in the infrared and this adiation can be used as a diagnostic probe for the physical conditions in the interior of this egion (Hayakawa, 1988; Nakada, 1988).

### I. THEORETICAL FORMALISM AND NUMERICAL RESULTS

### II.1 The Electric Dipole Transitions

Several laboratory studies have shown that the fundamental electronic state of  $OH^+$  on is  $X^3\Sigma^-$  (see for instance Merer et al., 1975). For a molecule with state  $X^3\Sigma^-$ , the electron rbital angular momentum in the nuclear axis is zero ( $\Lambda=0$ ) and the electron spin interactions ith the molecular rotation obey the following expressions of rotational energy:  $F_1(J=N+1)$ ,  $F_2(J=N)$  and  $F_3(J=N-1)$  (Dixon, 1959), were N is the rotational quantum number, excluding the lectron spin and J is the total rotational quantum number including the molecular rotation with he electron spin S (J=|N+S| to |N-S|).

For electric dipole transitions in state  $X^3\Sigma^-$  (see Figure 1), with coupling belonging to

case b, the selection rules are  $\Delta N = \pm 1$  and  $\Delta J = 0, \pm 1$  (Somerville, 1978) and the emissions of  $OH^+$  (N=1 $\rightarrow$ 0) in level v=0 may occur in (Bekooy et al., 1985):

$$v_1 = F_3'(0) - F_1''(1) = 909121.07 \text{ MHz}$$

$$v_2 = F_1'(2) - F_1''(1) = 971811.77 \text{ MHz}$$

$$v_3 = F_2'(1) - F_1''(1) = 1033078.37 \text{ MHz}$$
(1)

The line strength (Hon1 - London factors) of these rotational transitions are giving by (Somerville, 1978):

$$S_{J}(J' = 0 + J'' = 1) = \frac{1}{J(2J - 1)(2J + 1)}$$

$$S_{J}(J' = 2 + J'' = 1) = \frac{J(2J + 3)}{2J + 1}$$

$$S_{J}(J' = 1 + J'' = 1) = \frac{1}{J}$$
(2)

which for J=1 assume values 1/2, 5/3 and 1 for  $\nu_1$ ,  $\nu_2$  and  $\nu_3$ , respectively. This clearly shows that  $\nu_2$  corresponds to the strongest rotational transition between the N=1  $\rightarrow$  0 transitions in the v=0 level of the X<sup>3</sup> $\Sigma^-$  state of OH<sup>+</sup> ion and holds  $\sim$  56% of N=1  $\rightarrow$  0 total intensity. It should be noted that these three rotational transitions are localized in the wavelength range of 290  $\mu m \, \text{--}$ 330 µm and consequently in the SMM region of the electromagnetic spectrum.

The Einstein transition probability for spontaneous emission of a rotational line is given by (Somerville, 1978):

$$A(J+1 \to J) = 1.16395 \times 10^{-38} \text{ v}^3 \text{ } \mu^2 \frac{s_J}{s_2} \qquad (s^{-1})$$
 (3)

where  $\nu$  is the rotational frequency in Hz,  $\mu$  = 2.32 Debye (Werner et al., 1983) is the electric dipole moment for  $OH^{+}$ ,  $S_{J}$  is the corresponding line strength for V and  $S_{2} = 2J+1$  is the statistical weight of the upper level.

The emission in FIR is optically thin. For an optically thin line, the emission

intensity in the FIR or SMM regions is given by (Watson, 1985a; Melnick, 1986; Sternberg, 1988):

$$I_{j \to i} = \frac{h}{4\pi} v_{ij} A_{ji} f_{j} n_{ion} d\ell,$$

or

$$I_{j \to i} = \frac{h}{4\pi} v_{ij} A_{ji} f_{j} [n_{ion} \ell] (erg s^{-1} cm^{-2} sr^{-1})$$
 (4)

where h is Planck's constant,  $v_{ij}$  is the frequency of the observed transition,  $A_{ji}$  is the Einstein transition probability for spontaneous emission,  $f_{ij}$  is the fraction of all the ions in the upper level j,  $n_{ion}$  is the volumetric density of the ionic species,  $\ell$  is the length of the emitting region along the line of sight and the product  $[n_{ion}.\ell] \equiv [N] = cm^{-2}$ . For a kinetic temperature  $T_K = 70K$  which is appropriate for Orion-KL we have a fractional population  $f_{ij}$  (Herzberg, 1950) of OH ions in different rotational levels N=0,1 and 2 for v=0 level of the  $X^3\Sigma^-$  state as  $f_{ij} = 0.338$ ,  $f_{ij} = 0.516$  and  $f_{ij} = 0.223$ . This clearly shows that for  $T_K = 70K$  the maximum OH fractional population is in N=1 level and OH+(N=1 \to 0) transition occur in emission and with  $N_{OH} = 1.8 \times 10^{13}$  cm<sup>-2</sup> produced by a dissociative shock propagating with a velocity  $v_{ij} = 60 \, \mathrm{km \, s^{-1}}$  in a dense gas cloud of density  $n=10^5$  cm<sup>-3</sup> (Neufeld and Dalgarno, 1989) results in the following intensity values. following intensity values.

TABLE 1 - Values for N=1→0 in level v=0 electric dipole moment transitions of  $OH^+(X^3\Sigma^-)$  ion in Orion-KL

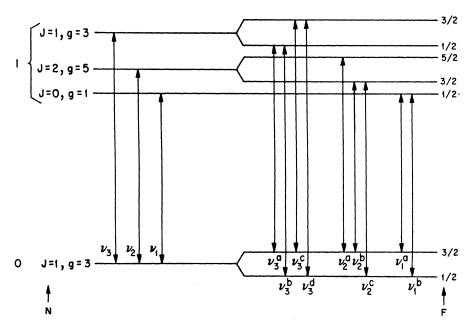
ν(MHz)	J'-J"	s <sub>j'j'</sub>	g <sub>2</sub> =2J+1	$A \times 10^{-2} (s^{-1})$	Ix10 <sup>-5</sup> (cgs)	
909121.07	0 – 1	0.333	1	1.569	6.986	
971811.77	2 – 1	1.667	5	1,917	9.124	
1033078.37	1 – 0	1.000	3	2.302	11.647	

# II.2 The Hyperfine Rotational Structure

The nuclear spin of  $OH^{+}$  ion is 1/2, due to the hydrogen nucleus ( $\vec{I}=1/2$ ). Thus the stal angular momentum F of the molecule is given by

$$F = J+I \tag{5}$$

here J is the rotational quantum number including the electron spin and F can assume values rom |J+I| to |J-I|. Due to the nuclear spin, each level of spin-rotation will be split in everal components. These hyperfine transitions are shown in Figure 1.



igure 1 - Hyperfine transitions for N=1  $\rightarrow$  0 in level v=0 of  $^{16}0^{1}\text{H}^{+}$  ion in state  $X^{3}\Sigma^{-}$ .

For the electric dipole transitions such as  $v_2(N'=1, J'=2 \rightarrow N''=0, J''=1)$ , we will have be corresponding hyperfine transitions  $v_2$ ,  $v_2^b$  and  $v_2^c$  with  $(\Delta F=0,\pm 1)$ . The frequencies of these perfine transitions (Table 2) were measured in laboratory with great accuracy by Bekooy et al. 975)

$$v_2^a(J'=2, F'=5/2 \rightarrow J''=1, F''=3/2) = 971803.8 \text{ MHz}$$

$$v_2^b(J'=2, F'=3/2 \rightarrow J''=1, F''=3/2) = 971919.2 \text{ MHz}$$

$$v_2^c(J'=2, F'=3/2 \rightarrow J''=1, F''=1/2) = 971805.3 \text{ MHz}$$
(6)

The HUnl-London factors for these hyperfine transitions can be determined through's following formulas, adopted from Condon and Shortley (1959):

$$[(J'=2,F'=5/2) - (J''=1,F''=3/2)] = \frac{(F+J-I-1)(F+J-I)(I+J+F+1)(I+J+F)}{4F} S_{J}(J'=2 \rightarrow J''=1)$$

$$[(J'=2,F'=3/2) - (J''=1,F''=3/2)] = \frac{(2F+1)(F+I-J)(F+I-J+1)(I+J+F)(I+J+F)}{4J(F+1)} S_{J}(J'=2 \rightarrow J''=1)$$

$$[(J'=2,F'=3/2) - (J''=1,F''=1/2)] = \frac{(F+J-I-1)(F+J-I)(I+J+F+1)(I+J+F)}{4F} S_{J}(J'=2 \rightarrow J'=1)$$

$$(7)$$

where  $S_J(J'=2+J''=1)=5/2$  (Eq.(2)) is the Hönl-London factor corresponding to  $v_2=F_1'(2)-F_1''(1)=971811,77$  MHz. Giving adequate values to F, J and I one will find the following values of Hönl-London factors for the hyperfine transitions  $v_2^a$ ,  $v_2^b$ ,  $v_2^c$ :3, 1/3 and 5/3, respectively. The Hönl-London factors for the other hyperfine transitions corresponding to  $v_1$  and  $v_2$  were evaluated in a similar way (see Condon and Shortley, 1959) and its values are given in Table 2. The Hönl-London factors corresponding to  $v_2$  shows that the line  $v_2^a$  is the strongest between the three hyperfine transitions coming from  $v_1$  and  $v_2$  and all of them (as well as the other hyperfine transitions coming from  $v_1$  and  $v_2$ ) fall in the SMM region of the electromagnetic spectrum since their wavelengths are localized also in the range 290  $\mu$ m - 330  $\mu$ m.

It was said above that the probability for a rotational transition  $A(J+1\to J)$  is given by Eq. (3) and now with  $g_2=2F'+1$  the emission intensities I (Eq.(4)) were evaluated for these hyperfine transitions and their values are shown in Table 2.

TABLE 2 - Values of hyperfine transitions for N=1+0 in level v=0 in state  $X^3\Sigma^-$  of OH<sup>+</sup> ion.

Frequency (MHz)	N',J'←N'',J''	F' ← F"	ν	g <sub>2</sub> =2F'+1	S <sub>J'J"</sub>	$A \times 10^{-2} (s^{-1})$	I x 10 <sup>-5</sup> (cgs)
909 045,2	1,0 - 1,0	1/2-1/2	ν <mark>b</mark>	2	0.333	0.785	3.495
909 158.8	1,0 - 1,0	1/2-3/2	ν <mark>á</mark> 1	2	0.667	1.569	6.986
971 803.8	1,2 - 1,0	5/2-3/2	$v_2^{\dot{a}}$	6	3.000	2.875	13.683
971 805.3	1,2 - 1,0	3/2-1/2	$v_2^{c}$	4 .	1.666	2.395	11.399
971 919,2	1,2 - 1,0	3/2-3/2	$v_2^{\bar{b}}$	4	0.334	0.479	2.280
1 032 997,9	1,1 - 1,0	1/2-1/2	ν <sub>3</sub> ̄̄̄̄̄̄	2	0.667	2 <b>.3</b> 02	11.646
1 033 004.4	1,1 - 1,0	3/2-1/2	l à	4	0.333	0.575	2.909
1 033 111,8	1,1 - 1,0	1/2-3/2	ν <sub>3</sub> a	2	0,333	1.151	5.824
1 033 118,6	1,1 - 1,0	3/2-3/2	ν <mark>c</mark> γ3	4	1.667	2.879	14.567

### III. DISCUSSION

Fast dissociative shocks in molecular clouds can produce detectable radio, optical and infrared recombination lines (McKee and Hollenbach, 1987). The FIR and SMM regions are not explored in an extension comparable to the radio region so far (see for instance Melnick, 1986) Observations in this part of the spectrum is important to interstellar gas-phase chemistry sinc many radicals and molecular ions have their fundamental transitions there.

The chemistry of Orion-KL region is governed mainly by molecular shock waves and consequently this fact can contribute, so that the abundance of a given molecular specie be enhanced by several times in this source due to high temperature chemistry, if compared with th rest of the Orion molecular cloud. The low temperature ion-molecule chemistry was discussed by Singh and Almeida (1980) and Almeida and Singh (1981). Recently, Neufeld and Dalgarno (1989) showed that the dissociative "J" type shocks (where the magnetic field is not important and the molecule is dissociated by a shock with velocity  $v_s \ge 50 \text{ km s}^{-1}$  (Hollenbach and McKee, 1979; Chernoff et al., 1982)), propagating in dense gas clouds can produce the OH ion in the hot post-shock cooling gas behind the fast dissociative "J" type shock with an abundance  $N\ge 1.5\times 10^{1.5}$  cm<sup>-2</sup> for a shock propagating with a velocity  $v_s=80 \text{ km s}^{-1}$  in a gas of density  $v_s=10^{-1} \text{ cm}^{-3}$ . Plus, between the various FIR and SMM molecular lines presently detected at  $\lambda < 350 \mu \text{m}$ , mainly in Orior -KL region, that require observations in airborne spectrometers (for instance NASA's Gerald P. Kuiper Airborne Observatory - KAO), we note the presence of the transitions in the hydrides: Para -  $H_3O^+:J_K=1_0 \to 0_0$  in 304.5  $\mu \text{m}$  and  $NH(X^3\Sigma^-):N=1 \to 0$  in 307.6  $\mu \text{m}$  (Watson, 1985b), which are localized also in the interval of 290  $\mu \text{m}$  -330  $\mu \text{m}$  of the OH+ ion SMM transitions.

localized also in the interval of 290 µm -330 µm of the OH+ ion SMM transitions. Thus, the indication seems to be that OH+ SMM emission may yield detectable with a gas kinetic temperature  $T_K \gtrsim 50 K$  with the largest column density N=3.0x10<sup>13</sup> cm<sup>-2</sup>, which corresponds to a gas density n=10<sup>6</sup> cm<sup>-3</sup> and a shock velocity  $v_s = 100 \text{ km s}^{-1}$ , though weak in the hot shocked gas region in Orion-KL (Neufeld and Dalgarno, 1989) that is, measured in the post shock material of the quiet and hot  $(v_{LSR} \sim 9 \text{ km s}^{-1}; \Delta v \sim 4 \text{ km s}^{-1})$  region (this sub-region wa previously called the compact "ridge") (Ohishi et al., 1986).

Future balloon-platform observatories and space experiments such as the NASA's Space Infrared Telescope Facility (SIRTF) with a 2-700  $\mu$ m multiband imaging photometer should be capable of detecting these emission lines, thus serving as an unambiguous indicator of high temperature or shock waves chemistry.

### ACKNOWLEDGEMENTS

The author is grateful to Prof. A. Dalgarno and Dr. P.D. Singh for stimulating discussions and encouragement during the progress of this work.

### REFERENCES

Melnick, G., 1986, in Astrochemistry, IAU Symposium no 120, (eds.) M.S. Vardya and S.P. Tarafdar, p.137. Viscuso, P.J., Stacey, G.J., Fuller, C.E., Kurtz, N.T. and Hartwit, M., 1985a, Ap.J., 296, 142. Viscuso, P.J., Stacey, G.J., Harwitt, M., Haas, M.R., Erickson, E.F. and Duffy, P.B., 1985b, Ap. J., 296, 149. Watson, D.M., Genzel, R., Townes, C.H. and Storey, J.W.V., 1985a, Ap.J., 298,316. Ziurys, L.M., 1988, Ap.J., 324, 544. Merer, A.J., Malm, D.N., Martin, R.W., Horani, M. and Rostas, J., 1975, Can. J. Phys., 53, 251. Dixon, R.N., 1959, Can. J. Phys., 37, 1171. Somerville, W.B., 1978, Atomic and Mol. Phys., 13, 383. Bekooy, J.P., Verhoeve, P. Meerts, W.L. and Dymanus, A., 1985, J. Chem. Phys., 82, 3868. Werner, H.-J., Rosmus, P. and Reinsch, E.-A., 1983, J. Chem. Phys., 79, 905. Hollenbach, D., and McKee, C.F. 1979, Ap. J. Suppl., 41, 555. Herzberg, G., 1950, Spectra of Diatomic Molecules, 2nd ed., Van Nostrand Co., Princeton, N.J., p. Neufeld, D.A. and Dalgarno, A., 1989, to appear in Ap. J. Condon, E.U. and Shortley, G.H., 1959, The Theory of Atomic Spectra, Cambridge University Watson, D.M., 1985b, Phys. Ser., T11, 33. Singh, P.D. and Almeida, A.A., 1980, Astron. Astrophys., 84, 177. Almeida, A.A. and Singh, P.D., 1981, Astron. Astrophys., 95, 383. Nakada, Y., 1988, Publ. Astron. Soc. Japan, 40, 331. Ohishi, M., Kaifu, N., Suzuki, H., and Morimoto, M., 1986, Astroph. Space Sc., 118, 405. Hayakawa, S., 1988, Astroph. Sp. Sc., 144, 629. Chernoff, D.F., Hollenbach, D.J., and McKee, C.F., 1982, Ap.J. (Letters), 259, L97.

McKee, C.F. and Hollenbach, D.J., 1987, Ap. J., 322, 275.

Sternberg, A., 1988, Ap. J., 332, 400.

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