TEMPERATURE GRADIENTS AND CHEMICAL ABUNDANCES IN H II REGIONS

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We have collected a sample of 112 H II regions with measurements of $T_{\rm e}[{\rm N\,II}]$ and $T_{\rm e}[{\rm O\,III}]$ to study the effect of using relations between these two temperatures in the determination of oxygen abundances. The results confirm that the temperature relation depends on the degree of ionization of the H II regions. We propose new temperature relations that consider this dependence and lead to better estimates of electron temperatures and oxygen abundances.

When the observed spectra of H II regions allow the measurement of the line intensity ratios used to estimate electron temperatures, their chemical abundances can be calculated with the direct method. A two-zone ionization structure is commonly assumed, with $T_{\rm e}[{\rm N\,II}]$ associated to the emission of the low ionization ions, like O⁺ or N⁺, and $T_{\rm e}[{\rm O\,III}]$ to the high ionization ions, like O⁺⁺. When only one of these temperatures is measured, a value for the other one can be derived using a temperature relation. In order to study the effect of temperature relations on the derived metallicities, we have compiled a sample of 112 H II regions with measurements of both $T_{\rm e}[{\rm N\,II}]$ and $T_{\rm e}[{\rm O\,III}]$.

We find that the $T_{\rm e}[{\rm N~II}]-T_{\rm e}[{\rm O~III}]$ relation depends on the degree of ionization, confirming a previous result reported by Pilyugin (2007) for a smaller sample of H II regions. Hence, we divided our sample in two groups according to their value of the line intensity ratio $P = I([{\rm O~III}] \ \lambda\lambda4959, 5007)/(I([{\rm O~III}] \ \lambda\lambda4959, 5007) + I([{\rm O~III}] \ \lambda3727))$, which is related to the hardness of the ionizing radiation. Robust fits provide new temperature relations for the two groups. For P < 0.65we find:

$$T_{\rm e}[{\rm N\,II}] = 0.5 T_{\rm e}[{\rm O\,III}] + 4400 {\rm K}$$

$$T_{\rm e}[{\rm O\,III}] = 0.8 \, T_{\rm e}[{\rm N\,II}] + 1300 \, {\rm K},$$

and for P > 0.65:

$$T_{\rm e}[{\rm N\,II}] = 0.6 T_{\rm e}[{\rm O\,III}] + 4600 {\rm K}$$

$$T_{\rm e}[{\rm O~III}] = 1.1 T_{\rm e}[{\rm N~II}] - 1700 {\rm K}.$$

We have calculated the differences between the oxygen abundances derived using both electron temperatures and the abundances implied by either $T_{\rm e}[{\rm N\,II}]$ or $T_{\rm e}[{\rm O\,III}]$ with the corresponding temperature relation. We compare the results with those obtained with one of the most used temperature relations, that of Campbell et al. (1986). The mean and standard deviation of the logarithmic abundances are 0.03 ± 0.10 and 0.01 ± 0.12 when the relation of Campbell et al. (1986) is used to derive $T_{\rm e}[\rm N\,II]$ and $T_{\rm e}[{\rm O\,III}]$, respectively. Our new relations lead to small but consistent improvements on the differences: 0.02 ± 0.09 and -0.01 ± 0.10 . The fact that the improvements are not larger can be due to intrinsic dispersion in the dependence of the temperature relation with P or to errors in the line ratios used to derive electron temperatures. The complete analysis will be presented in Arellano-Córdova et al. (in prep.).

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